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## Key indicators

Single-crystal X-ray study

T = 90 K

Mean  $\sigma(\text{C}-\text{C}) = 0.001 \text{ \AA}$ 

R factor = 0.024

wR factor = 0.075

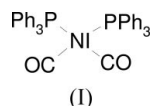
Data-to-parameter ratio = 83.4

For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.Dicarbonylbis(triphenylphosphine)nickel(0):  
a redetermination at 90 K

The structure of the title compound,  $[\text{Ni}(\text{C}_{18}\text{H}_{15}\text{P})_2(\text{CO})_2]$ , has been redetermined from high-resolution data collected at 90 K. The crystal structure reveals tetrahedral coordination geometry for nickel, with two triphenylphosphine and two carbonyl ligands. The Ni atom lies on a twofold rotation axis.

## Comment

The crystal structure of the title compound, (I), has been published previously (Krüger & Tsay, 1974). In the present study, data were collected at low temperature, using a diffractometer equipped with an APEX2 CCD area detector.



As shown in Fig. 1, the structure of (I) is composed of neutral  $[\text{Ni}(\text{PPh}_3)_2(\text{CO})_2]$  complex molecules. The results show that the coordination around the  $\text{Ni}^0$  atom, which lies on a twofold rotation axis, is distorted tetrahedral, composed of two triphenylphosphine P atoms [ $\text{Ni}-\text{P1} = 2.2167(1) \text{ \AA}$ ] and two carbonyl C atoms [ $\text{Ni}-\text{C19} = 1.7808(4) \text{ \AA}$ ]. The  $\text{Ni}^0$  atom assumes a distorted tetrahedral configuration as in a series of other complexes:  $[\text{Ni}(\text{R}_3\text{P})_2(\text{CO})_2]$ , where  $\text{R}_3\text{P}$  is tricyclohexylphosphine (Del Pra *et al.*, 1981), diphenyl(2-pyridinyl)phosphine (Wang *et al.*, 1989), tris(*o*-tolylxy)phosphine

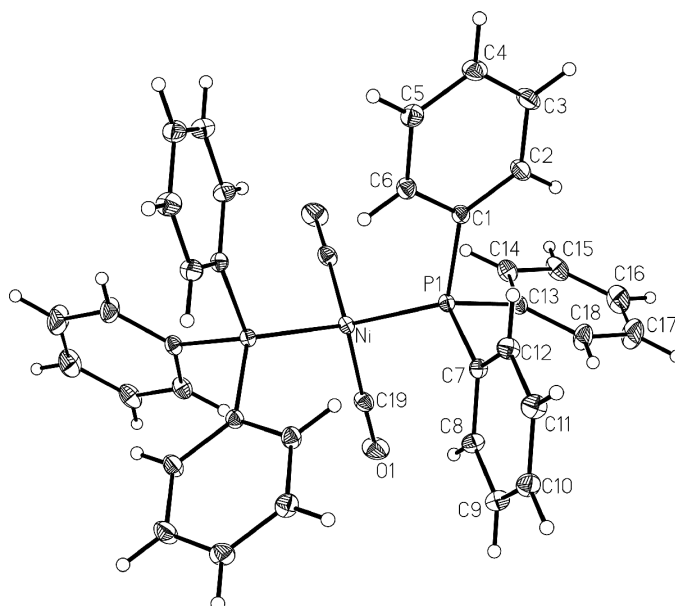


Figure 1  
View of (I) (50% probability displacement ellipsoids).

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(Meichel *et al.*, 2002), (4-hydroxybutyl)diphenylphosphine (Reinhard *et al.*, 2003) and 1,3,5-triaza-7-phosphaadamantane (Darensbourg *et al.*, 1999). The geometric parameters of the present structure agree well with those previously reported by Krüger & Tsay (1974), but with significantly improved precision. The precision of the Ni–P1 bond length (0.0001 Å), as well as the mean C–C bond length precision (0.0006 Å) for the low-temperature study, are better than the room-temperature data (0.001 and 0.004 Å, respectively). Similarly, the mean standard uncertainty for the C–C–C bond angles for the low-temperature data is 0.04°, whereas for the room-temperature data the mean standard deviation for the C–C–C bond angles is 0.2°.

## Experimental

The title compound was obtained from a commercial source (Aldrich Ltd). Crystals suitable for structure determination were grown from a solution in 1,2-dimethoxyethane in a dry-box.

### Crystal data

[Ni(C <sub>18</sub> H <sub>15</sub> P) <sub>2</sub> (CO) <sub>2</sub> ]	$D_x = 1.369 \text{ Mg m}^{-3}$
$M_r = 639.25$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 7626 reflections
$a = 11.7267(4) \text{ \AA}$	$\theta = 2.5\text{--}49.3^\circ$
$b = 8.1512(2) \text{ \AA}$	$\mu = 0.76 \text{ mm}^{-1}$
$c = 16.8433(5) \text{ \AA}$	$T = 90(1) \text{ K}$
$\beta = 105.612(1)^\circ$	Needle, colorless
$V = 1550.60(8) \text{ \AA}^3$	$0.20 \times 0.05 \times 0.04 \text{ mm}$
$Z = 2$	

### Data collection

Bruker SMART APEX2 diffractometer	16 260 independent reflections
$\omega$ scans	13 619 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.027$
$T_{\text{min}} = 0.863$ , $T_{\text{max}} = 0.970$	$\theta_{\text{max}} = 50.0^\circ$
99 172 measured reflections	$h = -25 \rightarrow 24$
	$k = -17 \rightarrow 17$
	$l = -36 \rightarrow 36$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0472P)^2 + 0.0217P]$
$R[F^2 > 2\sigma(F^2)] = 0.024$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.075$	$(\Delta/\sigma)_{\text{max}} = 0.002$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.76 \text{ e \AA}^{-3}$
16 260 reflections	$\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$
195 parameters	
H-atom parameters constrained	

**Table 1**

Selected geometric parameters (Å, °).

Ni–C19	1.7808 (4)	O1–C19	1.1486 (5)
Ni–P1	2.2167 (1)		
C19 <sup>i</sup> –Ni–C19	113.47 (3)	C19–Ni–P1	103.849 (14)
C19–Ni–P1 <sup>i</sup>	109.58 (1)	P1 <sup>i</sup> –Ni–P1	116.842 (6)

Symmetry code: (i)  $1 - x, y, \frac{3}{2} - z$ .

All H atoms were placed in calculated positions (C–H = 0.95 Å) and refined with a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: APEX2 (Bruker, 2004); cell refinement: APEX2; data reduction: SAINT-Plus (Bruker, 2004); program(s) used to solve structure: SIR97 (Altomare *et al.*, 1999); program(s) used to refine structure: XL in SHELXTL (Sheldrick, 2000); molecular graphics: XP in SHELXTL; software used to prepare material for publication: enCIFer (Allen *et al.*, 2004).

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